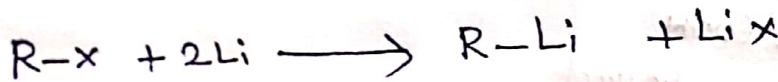


# Organolithium Compounds:

- The Organolithium Compounds are characterized by a C-Li bond.
  - The Organolithium Compounds, shows similar reactivity as Grignard reagents.
  - These are more reactive than the Grignard reagents.
  - Lithium is less electronegative than Carbon, and the C-Li bond is polarized  $(C^{\delta-}-Li^{\delta+})$
- 
- This C-Li act as both nucleophile and a base.

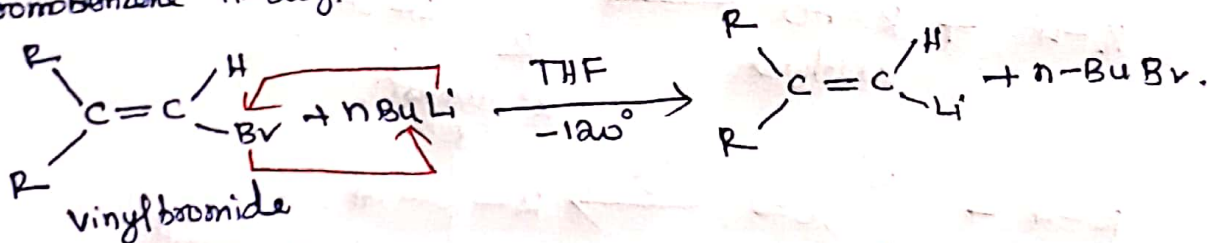
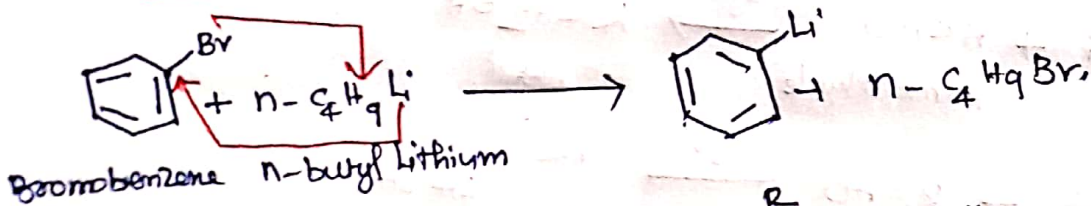
## Preparation:

(i) From alkyl (or) aryl halides:

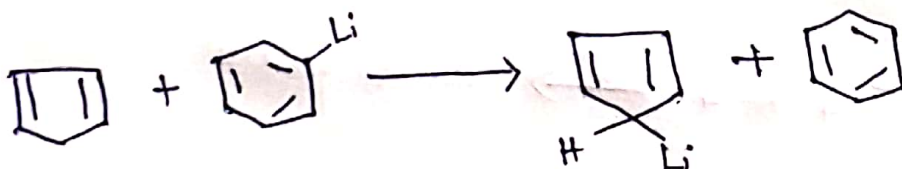


Where R = alkyl or aryl.

(ii) By halogen-metal exchange method:



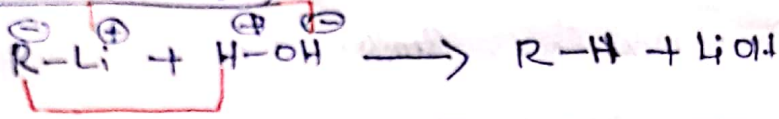
(iii) By metalation



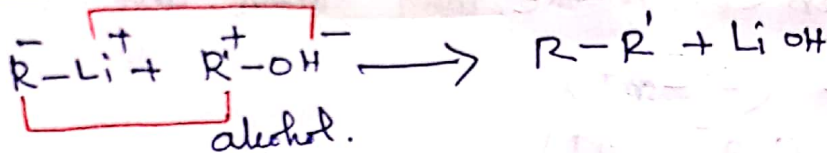
## Chemical Properties of C-Li (Reaction)

### (1) Reaction with compound containing active hydrogen

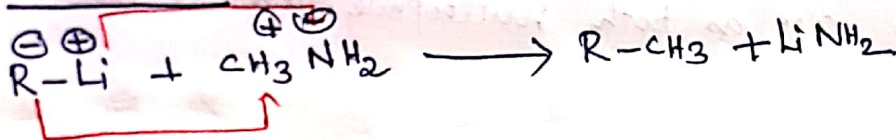
#### (i) With water (H<sub>2</sub>O)



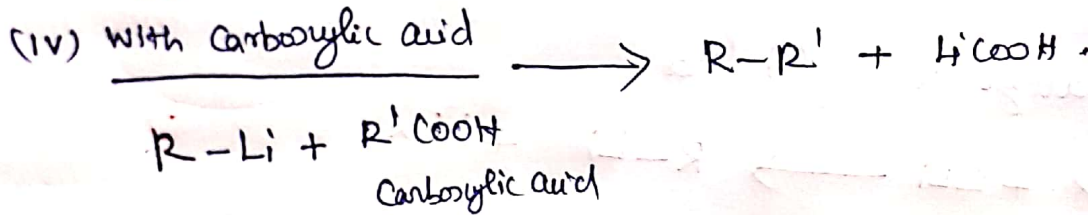
#### (ii) R-OH



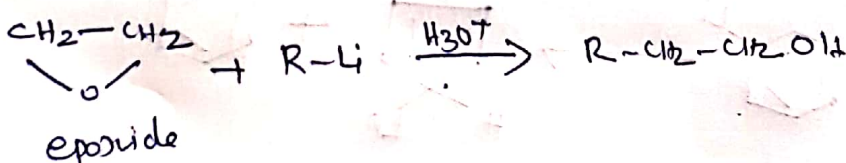
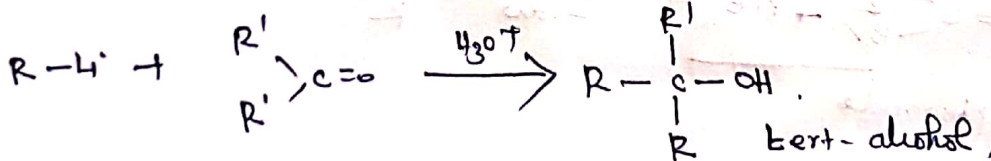
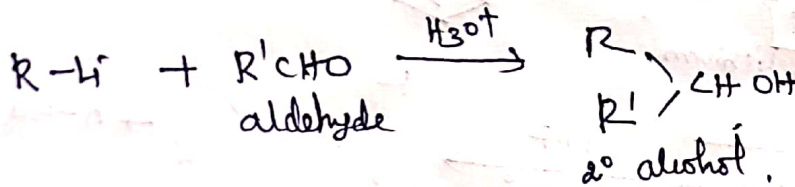
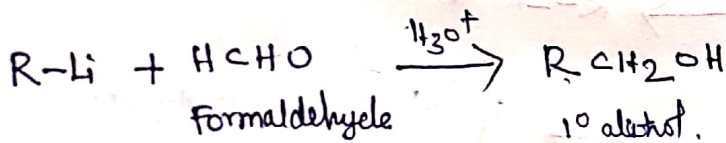
#### (iii) With amine



#### (iv) With carboxylic acid

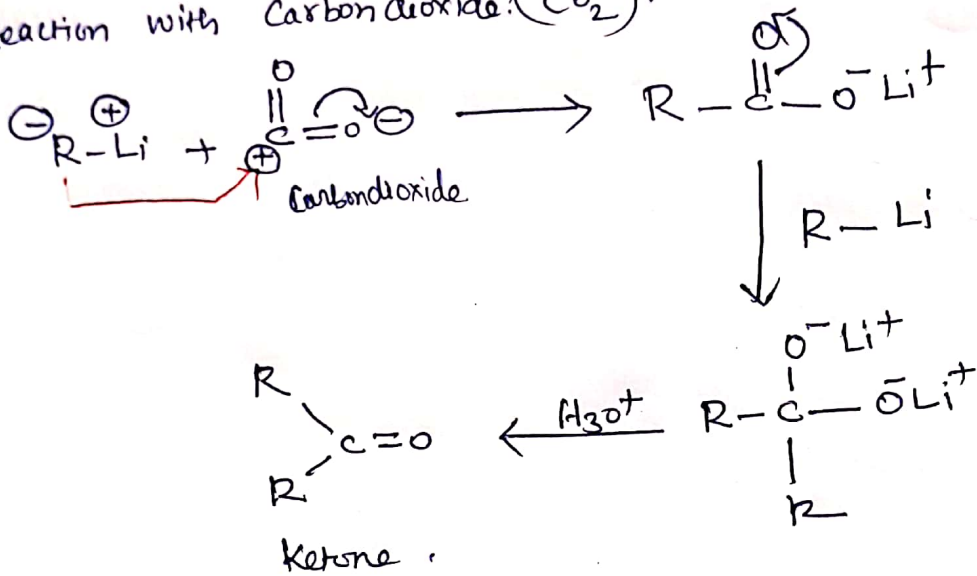


### (2) Reaction with Carbonyl compounds: (Aldehyde and ketone)

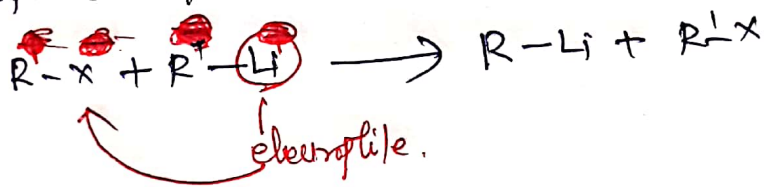


③

(B) Reaction with Carbon dioxide ( $\text{CO}_2$ ).



(4) Electrophilic displacement.



(5) Nucleophilic displacement.

