



**BHARATHIDASAN UNIVERSITY**

**Tiruchirappalli – 620024,  
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**Programme: M.Sc., Botany**

**Course Title : CELL BIOLOGY AND BIOINSTRUMENTATION**

**Course Code : 22PGBOT104**

**Unit – V**

**SEPARATION TECHNIQUES**

**Topic: Isoelectric Focusing**

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# ISOELECTRIC FOCUSING

## Isoelectric Focusing

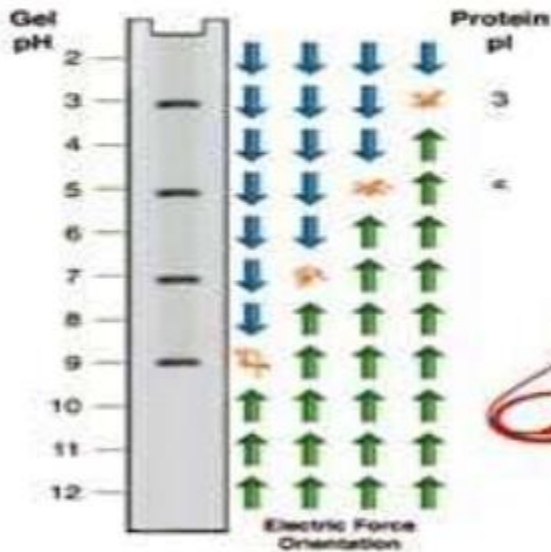
- Technique combining ideas of isoelectric points and electric fields
- Very high resolution technique for protein



# ISOELECTRIC FOCUSING

## ISOELECTRIC

## FOCUSING



# INTRODUCTION

- Proteins are separated in a pH gradient according to their isoelectric points.
- Basic principle involved is electrophoresis.
- Proteins are subjected to electric field in a pH gradient.
- Requires a solid surface normally Polyacrilamide.

# ISOELECTRIC POINT (pI)

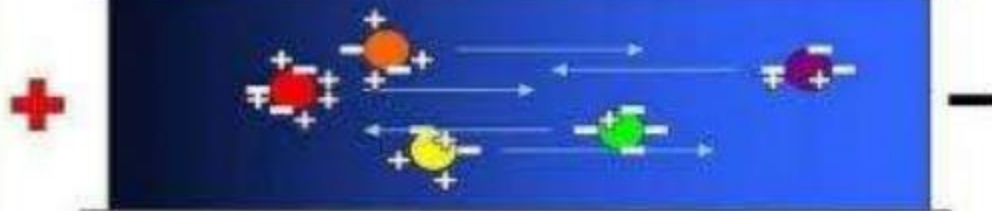
- The pH at which net charge on protein becomes zero.
  1. Below pI – Positive charge.
  2. Above pI – Negative charge.
- Proteins move toward the electrode with the opposite charge.
- During motion , proteins will either pick or loose protons.
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## Stable pH gradient

0 1 2 3 4 5 6 7 8 9 10 11 12 13 14



At low pH, most proteins have a positive charge while at high pH, most proteins have a negative charge.



When an electric field is present, the cathode and anode ends pull the proteins to their isoelectric point where each individual protein possesses a neutral charge.



The proteins stopped migrating because they've reached their isoelectric point at a unique pH level.

Sample applied at pH region below isoelectric points

Sample components migrate as Cations

Sample components are focused at their isoelectric points

Sample components migrate as anions

Sample applied at pH region above isoelectric points

Anode

$A^+$   $B^+$   $C^+$

$A^+$

$B^+$

$C^+$

$A^-$

$B^-$

$C^-$

$A^-$   $B^-$   $C^-$

Cathode

Lower pH

Increasing pH

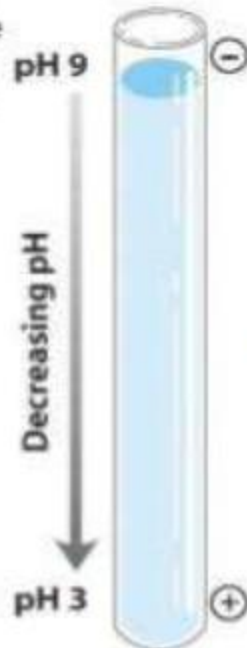
Higher pH

# PRINCIPLE

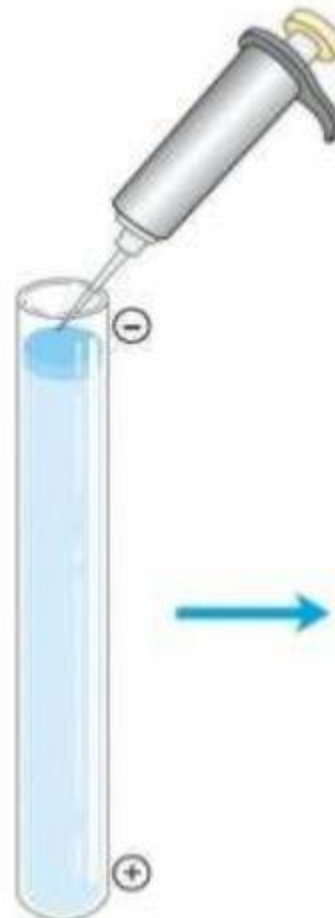
- All proteins have an isoelectric point pH.
- A procedure to determine the isoelectric point of proteins thus , a mixture of proteins can be electrophorised through a solution having a state pH gradient in form the anode to the cathode and a each protein will migrate to the position in the pH gradient according to its isoelectric point.This is called ISOELECTRIC FOCUSING.
- Protein migrate into the point where its net charge is zero-isoelectric pH.

# Isoelectric focusing

An ampholyte solution is incorporated into a gel.



A stable pH gradient is established in the gel after application of an electric field.



Protein solution is added and electric field is reapplied.



After staining, proteins are shown to be distributed along pH gradient according to their pI values.

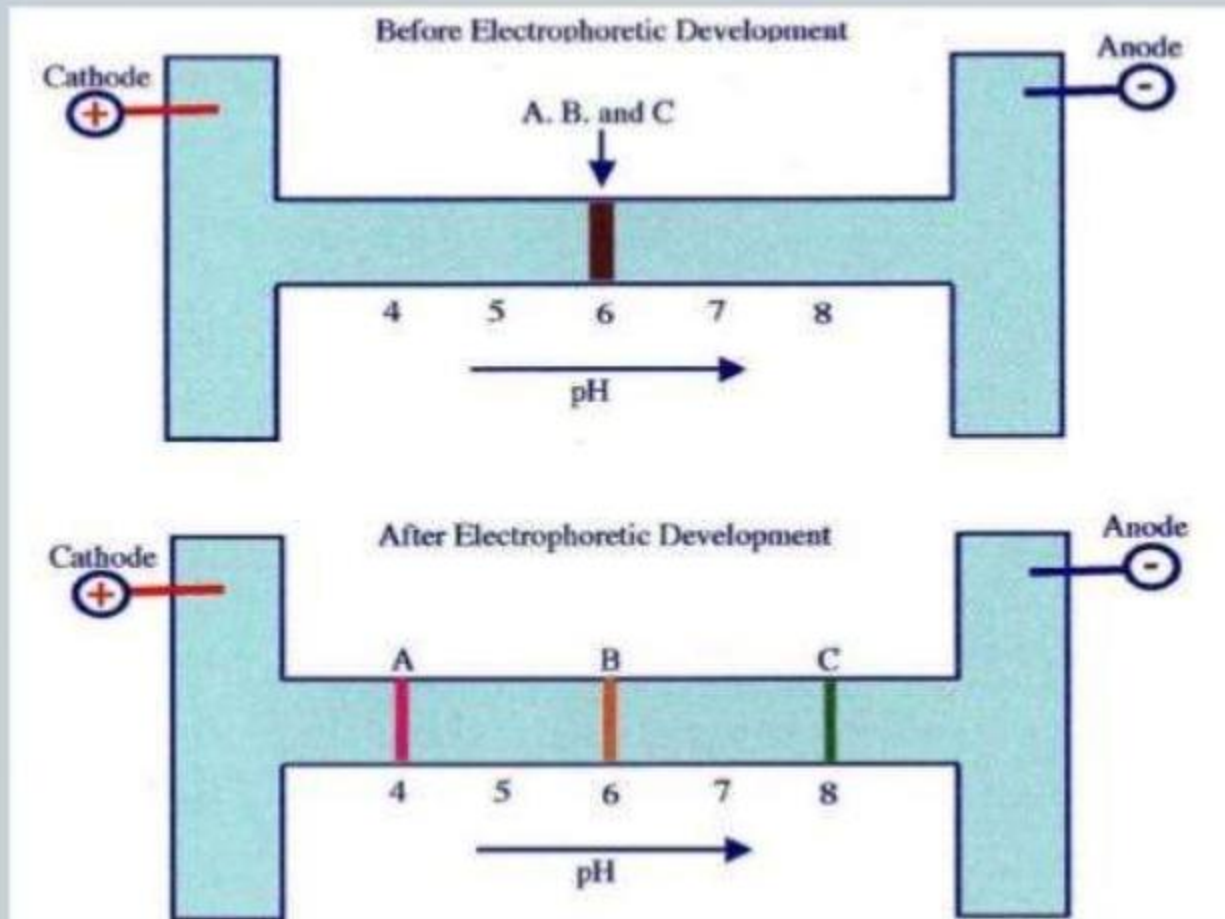


# CAPILLARY ISOELECTRIC FOCUSING

- pH gradient.
- Sample focusing and detection.
- Not useful for chiral compounds.
- Movement of gradient towards the detector.

# Capillary Isoelectric Focussing

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Theoretical separation of compounds with varying pI

# AMPHOLYTES

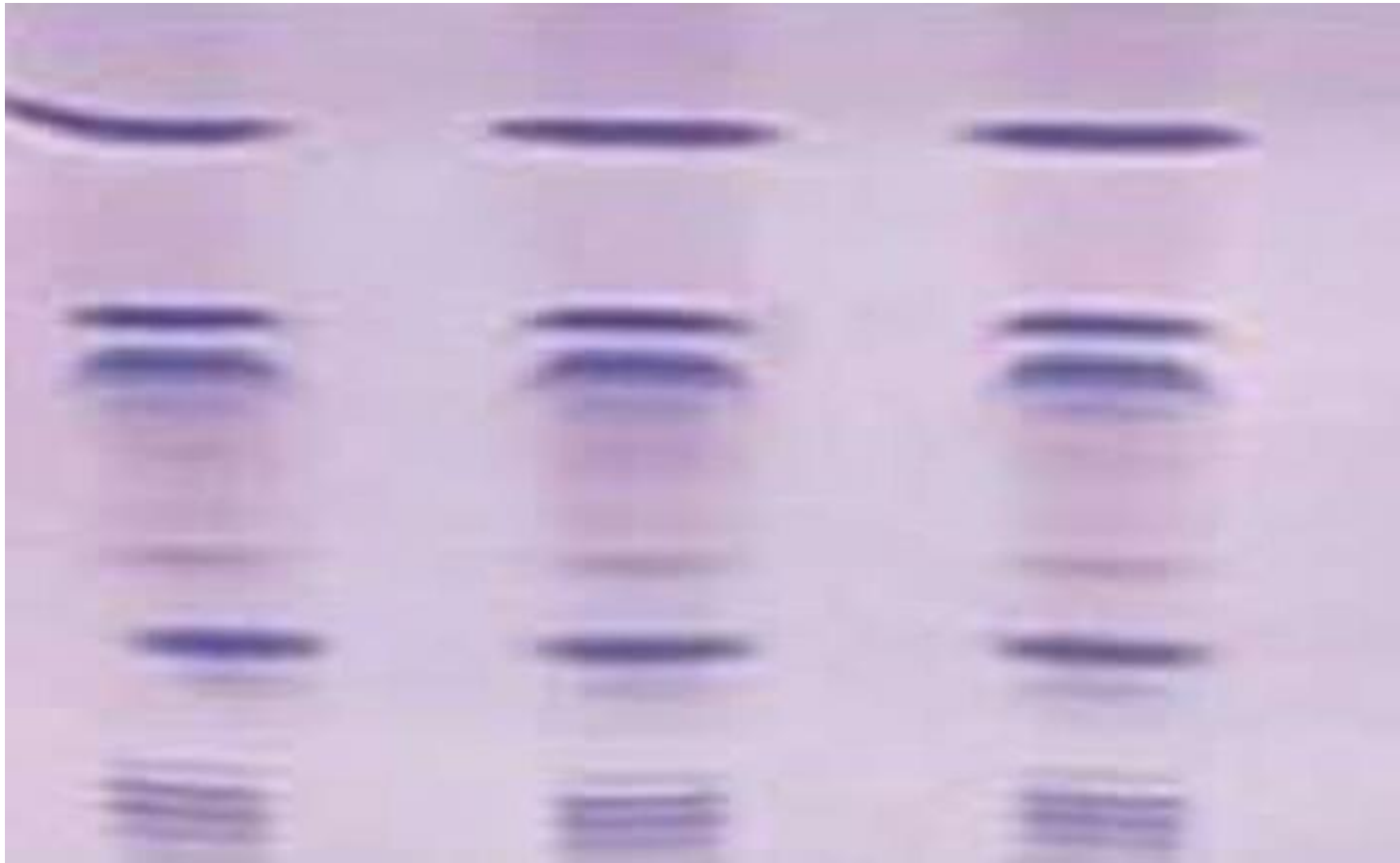
- Establishment of stable pH gradient is important.
- Achieved by means of commercially available synthetic carrier amphoteric electrolytes.
- 600-900 Da.
- Closely spaced pI and high conductivity.
- The curve is determined by pH interval covered by the ampholyte and the distance between electrodes.

# WORKING PROCEDURE

- Following chemicals are required-
  1. Acrylamide solution.
  2. Water.
  3. Ampholyte solution pH 3.5-10.
  4. Ampholyte solution pH 4-6.
- Add riboflavin and TEMED at the end.
- Remove bubbles.
- Fill the cassette completely with solution.
- Allow to polymerize at room temperature.

Run the gel at 150v for 30 min and then at 200v for 2 hr.

# ISOELECTRIC FOCUSING



# SET UP GEL

- Remove the comb carefully after gel has polymerized.
- Attach gel to the electrophoresis tank according the instructions of manufacturer.
- Add catholyte (sodium hydroxide) to the upper buffer chamber and anolyte (phosphoric acid) to the lower buffer chamber.

# SAMPLE PREPARATION AND LOADING

- Mix protein sample with equal volume of 2x loading buffer.
- Loading buffer includes the following reagents-

Ampholyte solution pH 3.5-10.

. Ampholyte solution pH 4-6.

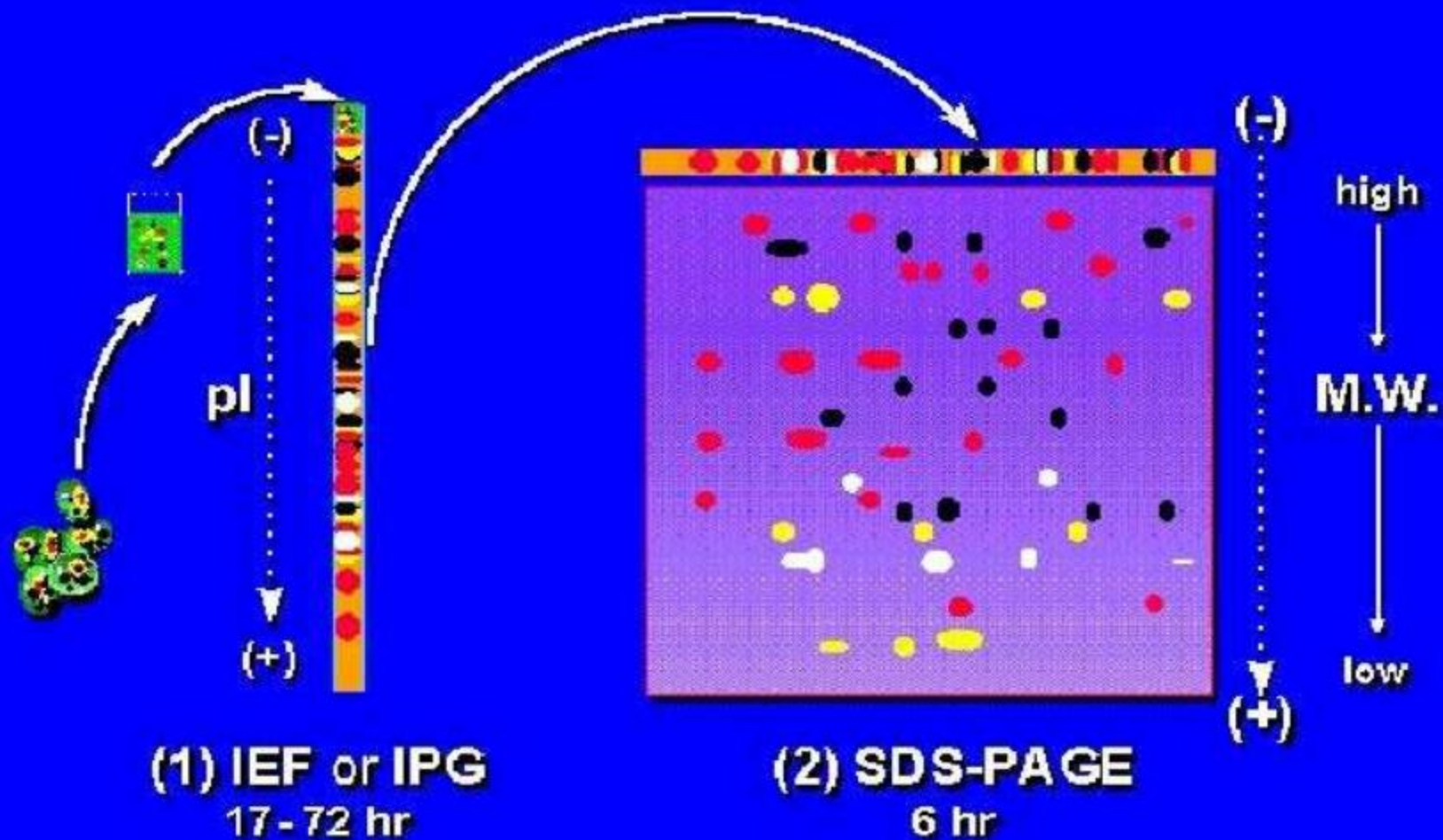
. 1% bromophenol blue.



# 2D GEL ELECTROPHORESIS

- Technique of IEF and SDS PAGE combined.
- Protein separated in two dimensions.
  1. On the base of  $pI$ .
  2. On the basis of molecular weight in normal SDS PAGE.
- Procedure can be adapted by combining IEF and PAGE.
- Series of spots formed in gel.

# Two Dimensional Electrophoresis



# ADVANTAGES

- Proteins that by as little as 0.001 pH units can be separated.
- As spreading of bands is minimized due to application of the applied field and the pH gradient, high resolution can be achieved.
- Isoelectric focusing (IEF) is a powerful analytical tool for the separation of proteins.
- Performing IEF is easier because the placement of sample application is not important

# DISADVANTAGES

- Carrier ampholytes are generally used in high concentration , a high voltage (upto 2000v) is necessary. As a result the electrophoretic matrix must be cooled which sometimes makes it difficult.
- Limited stability of solutions.
- Lot-to-lot inconsistency.
- Inadequate purity for application as a standard.

# APPLICATIONS

- For separating proteins and peptides.
- Used in limit test when the density of band is compared with the density of band of std prep.
- For research in Taxonomy , Cytology and Immunology etc.
- IEF gel is used as identity test when migration pattern on gel is compared with std preparation.
- Isoelectric focusing (IEF) offers an effective alternative to conventional electrophoresis for genetic marker typing.