COURSE TITLE: Biomolecules and Microbial Metabolism

Course Code: 24MICCC2

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PROPIONIC ACID AND FORMIC ACID FERMENTATION

INTRODUCTION OF FERMENTATION:

- The word "Ferment" arises from Latin " Fervere" which means "to boil".
- Fermentation is the act in which organic molecules such as glucose convert into acids, gases or alcohol in an oxygen free environment.
- Fermentation is named after products like alcoholic fermentation, lactic acid fermentation, propionic acid fermentation and formic acid fermentation.
- Fermentation process requires assistance from microorganisms such as *Enterococcus, Streptococcus* and *Lactobacillus* to complete.

PROPIONIC ACID FERMENTATION

- Propionic acid is from the Greek words "Protos First" and "Pion – Fat".
- ✤ It is a naturally occurring carboxylic acid with chemical formula CH₃CH₂COOH.
- It is colourless, corrosive liquid with a pungent Odor and in the pure state.
- It is soluble liquid in water, at any proportion .And widely found in animals , plants and microorganisms.
- **USES** Wide range of industries in the manufacture of pesticides , perfumes & pharmaceuticals and used as preservatives in grain , feed and food processing. It inhibit Molds , Bacillus, Aerobacter and other microorganisms.

CHEMICAL AND PHYSICAL PROPERTIES OF PROPIONIC ACID :

- IUPAC Name
- ✤ Other Names
- ✤ Molecular formula –
- Molar mass
- Appearance
- Melting point
- Boiling point

- Propionic acid.
- Ethane carboxylic acid.
 - $C_3H_6O_2$
 - 74.08g/mol.
 - Colourless.
 - -21°C.
 - 141°C.

PROPIONIC ACID FERMENTATION BY *Propionibacterium*

Propionic acid , CH₃CH₂COOH is produced by numerous bacteria such as *Clostridium* and *Propionicbacterium*.

Choice of Microorganisms :

Propionibacteria are gram-positive , facultative anaerobic bacteria that have been granted "generally recognized as safe" that include Propionibacterium freudenreichii, Propionibacterium shermanii and Propionibacterium acidipropionici.

METABOLIC PATHWAY OF PROPIONIC ACID FERMENTATION:



METABOLISM OF PROPIONIC ACID FERMENTATION:

- Pyruvic acid is obtained when the Propionibacteria dissimilate glucose into pyruvate via EMP pathway.
- Then, pyruvic acid is converted into oxaloacetic acid which is further reduced to succinic acid in two steps by reversal of the TCA cycle.
- The two steps include conversion of Malate by Malate dehydrogenase and conversion of Malate into Fumarate by Fumarate hydratase. And then succinate is formed.
- Due to the TCA cycle reversal process, the succinyl-CoA is obtained.



METABOLISM OF PROPIONIC ACID FERMENTATION:

- Succinic acid receives CoA by an enzyme, CoA-transferase. Then, Succinyl-CoA produces methyl malonyl-CoA by the action of Vitamin B12 – linked enzyme methyl malonyl mutase, which further catalyses an intramolecular rearrangement by the process of decarboxylation.
- Methyl malonyl-CoA is converted to propionyl-CoA.
- In the last step, Propionic acid is yielded from propionyl-CoA.



APPLICATION OF PROPIONIC ACID

- Food preservative: Propionic acid is a common preservative in food, especially bakery products. It's also used to treat drinking water for livestock.
- Animal feed preservative: Propionic acid is used to preserve compound feed, grain, and corn in animal feed.
- Chemical intermediate: Propionic acid is used as a chemical intermediate in the production of many other products, including polymers, flavorings, pesticides, and pharmaceuticals.
- Thermoplastics: Propionic acid is used in the production of thermoplastics.
- Anti-arthritic drugs: Propionic acid is used in the production of anti-arthritic drugs.
- *** Perfumes:** Propionic acid is used in the **production** of perfumes.
- Herbicides: Propionic acid is used in the production of herbicides.

FORMIC ACID FERMENTATION

- Formic acid is the simplest carboxylic acid and systematically called Methanoic acid.
- It is colourless, fuming liquid with a pungent odour and irritates the mucous membranes and blisters the skin.
- The chemical formula of Formic acid is HCOOH or HCO₂H. It is miscible with water and most polar organic solvents, and is soluble in hydrocarbons.



CHEMICAL AND PHYSICAL PROPERTIES **OF FORMIC ACID :**

- ✤IUPAC Name
- Molecular formula –
- ✤ Molar mass
- ✤ Appearance
- Melting point
- Boiling point

- Formic acid.
- Other Names
 Methanoic acid.
 - CH_2O_2
 - 46.02g/mol.
 - Colourless liquid. -
 - 8.4°C. _
 - 100.8°C.

FORMIC ACID FERMENTATION BY ENTEROBACTERIACEAE

- Formic acid fermentation is a process used by certain microorganisms to generate energy under anaerobic (oxygen-free) conditions that include the Enterobacteriaceae.
- These microorganisms can ferment monosaccharides, disaccharides, polyalcohol, and, less frequently, polysaccharides, via the glycolytic pathway, producing lactic, formic, succinic and acetic acids, and ethanol .This fermentation is called Mixed acid fermentation .

***** Choice of Microorganisms :

Enterobacteriaceae are **rod-shaped, gramnegative**, facultatively anaerobic ,non-spore forming bacteria that ferment sugars by EMP pathway.

METABOLIC PATHWAY OF FORMIC ACID FERMENTATION:



PFL- Pyruvate Formate Lyase FDH-Formate Dehydrogenase FHL-Formate hydrogenlyase LDH-Lactate Dehydrogenase ALDH-Aldehyde Dehydrogenase ADH-Alcohol Dehydrogenase ACK-Acetate Kinase

METABOLISM OF FORMIC ACID FERMENTATION

- Formic acid is a obtained from fermentation of Enterobacteriaceae and referred as formic acid fermentation.
- Formic acid fermentation pathway involves the breakdown of carbohydrates, primarily glucose into pyruvate and then formic acid, lactic acid, Acetate and other products by Embden-Meyerhof-Parnas Pathway or Glycolysis.
- Glycolysis takes place in the cytoplasm of the cell



METABOLISM OF FORMIC ACID FERMENTATION

- ✤In Glycolysis/EMP Pathway, Glucose is split into two molecules of Pyruvate, also produces energy in the form of 2ATP.
- Each pyruvate is split into Formate (Formic acid) by the enzyme **Pyruvate formatelyase(PFL**) and then acetyl-CoA.
- Formate can be further converted into CO₂ and H₂ by Formate dehydrogenase.
- It is also referred as mixed acid fermentation because not only formic acid production but also lactic acid, acetic acid and alcohol production.



APPLICATION OF FORMIC ACID

- An ingredient of medicines for rheumatism or vasoconstricting medicines.
- An **additive** for animal feeds and fungicides
- An ingredient of cosmetics for body care and regeneration
- Formic acid shows anti-bacterial properties as a result formic acid is of very high use in the agriculture industry.
- It is very well known as a pesticide and is proven beneficial to prevent crops from being prone to attacks from different kinds of pests