

Nanophysics - unit 1

Dr. K. Jeganathan
Department of Physics
Bharathidasan University
Tiruchirappalli - 24

22PH324A PHYSICS OF NANOMATERIALS AND APPLICATIONS

Length Scale in Physics

17/12

Length Scale

The concept of length scale is an important phenomena of different dimension of object. It gives idea of size of matter.

~~For~~ Ex: How big an atomic nuclei? the size of cell? , planet? and a galaxy?

If you are concentrating on low dimensional structure, there are four important units of length or distance

1. Bohr radius of hydrogen atom
2. The Compton wavelength of e^-
3. Classical electron wavelength
4. Planck length.

1. Hydrogen atom:

Hydrogen atom consists of a proton around which electron orbits

at a radius of r under the influence of the attractive force the Coulomb's force prevail in this two particle system attracted by the electric field, So we can use uncertainty principle. As the charged particle whizzing around, the radius depend on the electron charge e , and m , mass of e^- .



$p = 1836$
 $\frac{m_e}{m_p}$

Further, as the electron is in a fuzzed out orbit because of quantum effect, it involve Planck's constant h . Bohr radius does not depend on the proton mass, as 1836 times heavier than the e^- mass.

this is the relativistic calculation, so we don't use, c - velocity of light.

$$r = \frac{h^2}{m_e^2}$$

length scale is proportional to an inverse mass scale!

Hydrogen atom gets bigger when

it is bigger, or

gets smaller when e^- charge is smaller (more attractive force).

$$a_0 = r = \frac{20 h^2}{\pi m e^2} = 0.53 \text{ \AA}$$

half an angstrom.

2. Compton Wavelength

Compton effect is the scattering

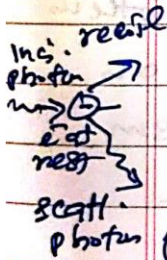
of photon by electrons. Compton

wavelength of a particle is ~~the~~

equal to the wavelength of a

photon whose energy is the same

(as the mass of the particle.



$$= \frac{h(1-\cos\theta)}{mc}$$

$$\lambda = \frac{h}{mc} = 4 \times 10^{-13} \text{ meters.}$$

(Pl. check $\frac{1}{2}$)

Again, marks scale sets the length scale.

$$h = 6.63 \times 10^{-34}$$
$$m_0 = 9.11 \times 10^{-31}$$
$$c = 3 \times 10^8 \text{ m}$$

$$\lambda = 0.2078 \text{ \AA}$$
$$= 6.124 \text{ \AA}$$

or $E = 1.02 \text{ MeV}$

3. The classical electron radius

The electromagnetic field produced by an electron, classically, has energy and thus should be counted as part of the mass of the electron! The length scale at which these effects become really important is called classical electron radius. It is not influenced by quantum field because only depend on the classical electromagnetism, which does not involve \hbar .

$$r_e = \frac{e^2}{mc^2} \\ = 3 \times 10^{-15} \text{ m.}$$

The classical e^- radius is $\frac{1}{137}$ as big as Compton wavelength of e^- .

4. Planck length

The Planck length is the smallest length at which space can be defined. At this scale the quantum gravity become important. The structure of spacetime becomes quite different from four dim. manifold.

Spacetime is like a bucket of foam or dust or bubbling sea of virtual black holes, in short, it is weird, but beyond that nobody really knows.

Planck length at which Q.M. gravity and relativity will interact very strongly \rightarrow Newton's gravitation - small const.

$$L_{\text{Planck}} = (h G / c^3)^{1/2}$$

$$= 1.6 \times 10^{-35} \text{ m}$$

classmate
Date _____
Page _____

Comparison of different length scale.

1 metre	10^0 m	Person
mm	10^{-3} m	
µm	10^{-6} m	blood cell
nm	10^{-9} m	
pm	10^{-12} m	atom
fm	10^{-15} m	
(9m) attometre	10^{-18} m	nucleus
zm - zeptometre	10^{-21} m	
ym - yocto meter	10^{-24} m	
	10^{-27} m	
	10^{-30} m	
	10^{-33} m	
	10^{-36} m	← Planck scale

